

Cambridge International Examinations

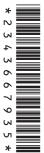
Cambridge International Advanced Subsidiary and Advanced Level

CHEMISTRY 9701/32

Paper 3 Advanced Practical Skills 2

May/June 2018

CONFIDENTIAL INSTRUCTIONS



Great care should be taken to ensure that any confidential information given does not reach the candidates either directly or indirectly.

The Supervisor's attention is drawn to the Supervisor's Report on page 7 which must be completed and returned with the scripts.

If you have any queries regarding these Confidential Instructions, please contact Cambridge stating the Centre number, the nature of the query and the syllabus number quoted above.

email info@cie.org.uk phone +44 1223 553554 fax +44 1223 553558

This document consists of 8 printed pages.



[Turn over

Safety

Supervisors are advised to remind candidates that **all** substances in the examination should be treated with caution.

Only those tests described in the Question Paper should be attempted.

In accordance with COSHH (Control of Substances Hazardous to Health) Regulations, operative in the UK, a hazard appraisal of the examination has been carried out. The following codes are used where relevant.

C corrosive
 HH health hazard
 F flammable
 MH moderate hazard
 T acutely toxic
 O oxidising

N hazardous to the aquatic environment

The attention of Supervisors is drawn to any local regulations relating to safety and first aid.

Hazard Data Sheets should be available from your chemical suppliers.

Before the examination

1 Access to the Question Paper is NOT permitted in advance of the examination.

2 Preparation of materials

Where quantities are specified for each candidate, they are sufficient for the experiments described in the Question Paper to be completed.

In preparing materials, the bulk quantity for each substance should be increased by 25% as spare material should be available to cover accidental loss. More material may be supplied if requested by candidates, without penalty.

All solutions should be bulked and mixed thoroughly before use to ensure uniformity.

Every effort should be made to keep the concentrations accurate.

If the concentrations differ slightly from those specified, the Examiners will make the necessary allowance. They should be informed of the exact concentrations.

3 Labelling of materials

Materials must be labelled as specified in these Confidential Instructions. Materials with an **FB** code number should be so labelled **without** the identities being included on the label. Where appropriate the identity of an **FB** coded chemical is given in the Question Paper itself.

4 Identity of materials

It should be noted that descriptions of materials given in the Question Paper may not correspond with the specifications in these Confidential Instructions. **The candidates must assume the descriptions given in the Question Paper.**

5 Size of group

In view of the difficulty of the preparation of large quantities of solution of uniform concentration, it is recommended that the maximum number of candidates per group be 30 and that separate supplies of solutions be prepared for each group.

Apparatus

- 1 In addition to the fittings ordinarily contained in a chemical laboratory, the apparatus and materials specified below will be necessary.
- 2 Pipette fillers (or equivalent safety devices), suitable eye protection and disposable gloves should be used where necessary.
- 3 For each candidate
 - $1 \times 250 \, \text{cm}^3 \, \text{beaker}$
 - $1 \times 100 \, \text{cm}^3 \, \text{beaker}$
 - 1 × 50 cm³ measuring cylinder
 - 1 × spatula
 - 2 × teat/dropping pipette
 - 1 × glass rod
 - 2 × crucible (capacity 15 cm³) with lid
 - 1 × heatproof mat
 - 1 × Bunsen burner
 - 1 × pipe-clay triangle
 - 1 × crucible tongs
 - 1 × tripod
 - 1 × gauze
 - $1 \times \text{stop watch (or sight of clock)}$
 - 2 × foamed plastic (polystyrene) cup
 - $1 \times \text{thermometer } (-10 \,^{\circ}\text{C to } +110 \,^{\circ}\text{C at } 1 \,^{\circ}\text{C})$
 - 8 × test-tube*
 - 1 × test-tube rack
 - 1 × wash bottle containing distilled water
 - 1 × pen for labelling glassware

paper towels

access to balance, weighing to a minimum of 0.1 g

Where balance provision is limited, some candidates should be instructed to start the examination with different questions. See the current syllabus for balance: candidate ratio.

^{*}Candidates are expected to rinse and reuse test-tubes where possible. Additional test-tubes should be available.

Chemicals required

- It is especially important that great care is taken that the confidential information given below does not reach the candidates either directly or indirectly.
- It should be noted that descriptions of substances given in the Question Paper may not correspond with the specifications in these Confidential Instructions. 2

3 Particular requirements

2.0–2.1g magnesium hydroxide 1.9–2.0g calcium hydroxide 1.4–1.5g calcium oxide 20 cm³ 1.0 mol dm⁻³ sulfuric acid 20 cm³ 0.2 mol dm⁻³ sulfuric acid 20 cm³ 0.2 mol dm⁻³ silver nitrate 20 cm³ 0.05 mol dm⁻³ sodium strips 0.5 mol dm⁻³ sodium carbonate	hazard	label	per candidate	identity	notes (hazards given in this column are for the raw materials)
FB 2 100 cm³ 3.0 mol dm⁻³ hydrochloric acid FB 3 1.9–2.0g calcium hydroxide FB 4 1.4–1.5g calcium hydroxide FB 5 20 cm³ 1.0 mol dm⁻³ sulfuric acid FB 6 20 cm³ 0.2 mol dm⁻³ barium chloride FB 7 20 cm³ 0.05 mol dm⁻³ silver nitrate magnesium ribbon 3 × 2 cm strips magnesium strips aqueous aqueous 10 cm³ 0.5 mol dm⁻³ sodium carbonate		FB 1	2.0–2.1g		$2.0-2.1\mathrm{g}$ of Mg(OH) $_2$ in a stoppered container.
FB 31.9–2.0gcalcium hydroxideFB 41.4–1.5gcalcium oxideFB 520 cm³1.0 mol dm⁻³ sulfuric acidFB 620 cm³0.2 mol dm⁻³ barium chlorideFB 720 cm³0.05 mol dm⁻³ silver nitratemagnesium ribbon strips3 × 2 cm stripsmagnesium stripsaqueous sodium carbonate10 cm³0.5 mol dm⁻³ sodium carbonate	[MH]	FB 2	100 cm ³		Dilute 272 cm³ of concentrated (35–37%; approximately 11 mol dm⁻³) hydrochloric acid [C][MH] to 1.00 dm³ with distilled water.
FB 41.4–1.5gcalcium oxideFB 520 cm³1.0 mol dm⁻³ sulfuric acidFB 620 cm³0.2 mol dm⁻³ barium chlorideFB 720 cm³0.05 mol dm⁻³ silver nitratemagnesium ribbon strips aqueous sodium carbonate3 × 2 cm aqueous carbonatemagnesium carbonate	[MH]	FB 3	1.9–2.0g	calcium hydroxide	$1.9-2.0\mathrm{g}$ of Ca(OH) $_2$ [C][MH] in a stoppered container.
FB 520 cm³1.0 mol dm⁻³ sulfuric acidFB 620 cm³0.2 mol dm⁻³ barium chlorideFB 720 cm³0.05 mol dm⁻³ silver nitratemagnesium ribbon strips3 × 2 cm strips aqueous 10 cm³magnesium strips 		FB 4	1.4–1.5g	calcium oxide	1.4–1.5g of CaO [C][MH] in a stoppered container.
FB 620 cm³0.2 mol dm⁻³ barium chlorideFB 720 cm³0.05 mol dm⁻³ silver nitratemagnesium ribbon sodium carbonate3 × 2 cm strips 10 cm³magnesium strips carbonate	[MH]	FB 5	20 cm ³	1.0 mol dm ⁻³ sulfuric acid	For preparation instructions see the current syllabus.
FB 7 20 cm³ 0.05 moldm⁻³ silver nitrate magnesium ribbon 3 × 2 cm strips magnesium strips aqueous sodium carbonate 10 cm³ 0.5 moldm⁻³ sodium carbonate		FB 6	20 cm³		Dissolve 48.8g of BaC I_2 .2H $_2$ O [T] in each dm 3 of solution. (Note: barium nitrate is not a suitable alternative.)
magnesium ribbon3 x 2 cm stripsmagnesium aqueousaqueous sodium carbonate10 cm³ carbonate0.5 mol dm⁻³ sodium carbonate		FB 7	20 cm³		For preparation instructions see the current syllabus. Each candidate will require 20 cm³ in a stoppered bottle. A darkened glass bottle should not be used.
10 cm³ carbonate	[F]	magnesium ribbon	3×2 cm strips	magnesium	Strips of Mg ribbon [F] cut into approximately 2 cm lengths.
		aqueous sodium carbonate	10 cm³		Dissolve 53.0g of Na ₂ CO ₃ [MH] in each dm³ of solution or dissolve 143.0g Na ₂ CO ₃ .10H ₂ O [MH] in each dm³ of solution.
dide 5 cm³ 0.5 mol dm⁻³ potassium iodide		aqueous potassium iodide	5 cm³	0.5 mol dm ⁻³ potassium iodide	Dissolve 83.0g of KI in each dm^3 of solution.

reagents. If necessary, they may be made available from a communal supply: however, the attention of the Invigilators should be drawn to the The reagents below should also be provided. Unless otherwise stated, each candidate should require no more than 10 cm³ of any of these fact that such an arrangement may lead to contamination of reagents and enhance the opportunity for malpractice between candidates.

hazard	label	notes
	dilute hydrochloric acid	
<u>5</u>	dilute nitric acid	
[MH]	dilute sulfuric acid	
[C][MH] [N]	aqueous ammonia	
<u></u>	aqueous sodium hydroxide	See identity details and preparation instructions in the current syllabus.
	0.1 moldm ⁻³ barium chloride or 0.1 moldm ⁻³ barium nitrate	
	0.05 moldm ⁻³ silver nitrate	
[MH]	limewater	
[MH]	acidified aqueous potassium manganate(VII)	

5 The following materials and apparatus should be available.

red and blue litmus papers, aluminium foil for testing nitrate/nitrite, wooden splints and the apparatus normally used in the Centre for use with limewater in testing for carbon dioxide

Responsibilities of the Supervisor during the examination

1 The Supervisor, or other competent chemist, must, out of sight of the candidates, carry out the experiments in Question 1 and Question 2 and complete tables of readings on a spare copy of the Question Paper. This should be labelled 'Supervisor's Results' and show the Centre number and appropriate session/laboratory number.

This should be done for **each session** held and **each laboratory** used in that session, and **each batch** of solutions supplied.

N.B. The Question Paper cover requests the candidate to fill in details of the examination session and the laboratory used for the examination.

It is essential that each packet of scripts contains a copy of the applicable Supervisor's Results as the candidates' work cannot be assessed accurately without such information.

2 The Supervisor must complete the Supervisor's Report on page 7 to show which candidates attended each session. If all candidates took the examination in one session, please indicate this on the Supervisor's Report. A copy of the Supervisor's Report must accompany each copy of the Supervisor's Results in order for the candidates' work to be assessed accurately.

The Supervisor must give details on page 8 of any particular difficulties experienced by a candidate, especially if the Examiner would be unable to discover this from the written answers.

After the examination

Each envelope returned to Cambridge must contain the following items.

- 1 The scripts of those candidates specified on the barcode label provided.
- 2 A copy of the Supervisor's Results relevant to the candidates in 1.
- **3** A copy of the Supervisor's Report, including details of any difficulties experienced by candidates (see pages 7 and 8).
- **4** The Attendance Register.
- **5** A Seating Plan for each session/laboratory.

Failure to provide appropriate documentation in each envelope may cause candidates to be penalised.

Colour blindness

With regard to colour blindness it is permissible to advise candidates who request assistance on colours of, for example, precipitates and solutions (especially titration end-points). Please include with the scripts a note of the candidate numbers of such candidates.

Experience suggests that candidates who are red/green colour-blind – the most common form – do not generally have significant difficulty. Reporting such cases with the scripts removes the need for a Special Consideration Form.

SUPERVISOR'S REPORT

This form must be completed and returned in t Results, the Attendance Register and the seati	
Centre number	Name of Centre
The candidate numbers of candidates attending ea	ach session were:
first session	second session

The Supervisor is required to give details overleaf of any difficulties experienced by particular candidates, giving names and candidate numbers. These should include reference to:

- any general difficulties encountered in preparation of materials;
- difficulties due to faulty apparatus or materials;
- accidents to apparatus or materials;
- assistance with respect to colour blindness.

Other cases of hardship, e.g. illness, temporary disability, should be reported directly to Cambridge on the Special Consideration Form.

Report on any difficulties experienced by candidates.
Declaration (to be signed by the Supervisor)
The preparation of this practical examination has been carried out so as to maintain fully the security of the examination.
Signed

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Name (in block capitals) (Supervisor)

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